Acta Crystallographica Section C

## Crystal Structure

Communications
ISSN 0108-2701

# catena-Poly[[dinitratocadmium(II)]bis[ $\mu$-bis(2-methyl-1 H-imidazol-1-yl)methane $\left.\left.-\kappa^{2} N^{3}: N^{3}\right]\right]$ 

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Received 7 August 2007
Accepted 5 September 2007
Online 22 September 2007

The title compound, $\left[\mathrm{Cd}\left(\mathrm{NO}_{3}\right)_{2}\left(\mathrm{C}_{9} \mathrm{H}_{12} \mathrm{~N}_{4}\right)_{2}\right]_{n}$, has a onedimensional double-bridged chain polymer structure with a 16 -membered macrometallacyclic tetragonal structural motif. The $\mathrm{Cd}^{\mathrm{II}}$ ion occupies a crystallographic inversion centre and is coordinated by four equatorial N atoms from four distinct bis(2-methylimidazol-1-yl)methane ligands and two apical nitrate O atoms to form a slightly distorted octahedral coordination geometry.

## Comment

Coordination polymers have attracted a great deal of attention because of their intriguing structural properties and potential applications in magnetism, electrical conductivity, ion exchange, separation and catalysis (Carlucci et al., 2003; Moulton \& Zaworotko, 2001). Studies in this field have been focused on the design and construction of novel coordination frameworks and on the examination of the relationships between their structures and properties. Many coordination polymers with diverse architectures, such as linear or zigzag chains, helices, honeycombs, square grids, ladders, brick walls and interwoven diamondoids, have been developed extensively (Bünzli \& Piguet, 2002; Sharma \& Rogers, 1999). More recently, flexible components, both ligands and metals, have been employed as versatile building blocks for coordination polymer construction, allowing access to topologies that are often unattainable using more rigid components (Jung et al., 2002). Bis(2-methylimidazol-1-yl)methane ( 2 -mBIM) is a bisimidazole ligand containing a connecting flexible methylene linker. We have reported an interesting infinite one-dimensional catenane polymer formed by the addition of a solution of $2-\mathrm{mBIM}$ to $\mathrm{AgNO}_{3}$ (Jin et al., 2006). We report here the interesting solid-state structure of a novel one-dimensional double-bridged chain polymer structure, (I), with a 16 -
membered macrometallacyclic tetragonal structural motif, formed by the addition of a solution of $2-\mathrm{mBIM}$ to $\mathrm{Cd}\left(\mathrm{NO}_{3}\right)_{2}$.

As revealed by single-crystal X-ray diffraction analysis, each $\mathrm{Cd}^{\mathrm{II}}$ cation occupies a crystallographic inversion centre and is coordinated by four N atoms from four distinct $2-\mathrm{mBIM}$ ligands and two O atoms from two nitrate anions to form a slightly distorted octahedral geometry about the cadmium center. The four N -donor atoms of the 2 -mBIM ligands lie in the equatorial plane and the two coordinated nitrate anions in the apical positions (Fig. 1). Selected $\mathrm{N}-\mathrm{Cd}-\mathrm{N}$ and $\mathrm{N}-\mathrm{Cd}-$ O bond angles are given in Table 1. The Cd1-N2 and Cd1$\mathrm{N} 5^{\mathrm{ii}}$ distances (see Table 1 for values and symmetry code) are within the range expected for such ligating bonds (Kasai et al., 2000). It is noteworthy that the nitrate anion, a monodentate ligand, has a $\mathrm{Cd} 1-\mathrm{O} 1$ distance that is slightly shorter than those reported for similar structures (Wang et al., 2000). The $\mathrm{Cd} 1 \cdots \mathrm{O} 2$ and $\mathrm{Cd} 1 \cdots \mathrm{O} 3$ distances are 3.319 (1) and 4.374 (2) Å, respectively.

(I)

Each pair of neighboring $\mathrm{Cd}^{\text {II }}$ centers is bridged by two 2 -mBIM ligands to form a 16 -membered macrometallacyclic tetragonal structural motif with a $\mathrm{Cd} \cdots \mathrm{Cd}$ separation of $9.654 \AA$. Each 2-mBIM ligand is coordinated to two $\mathrm{Cd}^{\mathrm{II}}$ cations and acts as a bridge ligand to form a one-dimensional double-bridged chain polymer structure that runs along the [ $01 \overline{1}$ ] direction. The conformation of the $2-\mathrm{mBIM}$ ligand in (I) is considerably different from that of free 2 -mBIM (Jin \& Gong, 2005), since the two methyl groups at the 2-position of the coordinated ligand point in the same direction (see Fig. 1). Each pair of 2 -mBIM ligands in the same 16 -membered metallacycle is centrosymmetric. The dihedral angle between the two imidazole rings of the same 2 -mBIM ligand is $59.6^{\circ}$. Significant $\pi-\pi$ contacts [the shortest $\mathrm{N} \cdots \mathrm{N}$ distance is 3.707 (2) $\AA$ ] and weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions between adjacent one-dimensional double-bridged chain polymers are


Figure 1
The structure of (I), showing the atom numbering. Displacement ellipsoids are drawn at the $50 \%$ probability level. H atoms have been omitted for clarity. [Symmetry codes: (i) $-x+1,-y,-z+2$; (ii) $x, y-1$, $z+1$; (iii) $-x+1,-y+1,-z+1$.]
present. Although similar one-dimensional double-bridged chain polymer structures have been documented for bipyridine ligands (Dong et al., 2000), coordination polymer structures containing $\mathrm{Cd}\left(\mathrm{NO}_{3}\right)_{2}$ and imidazole ligands are still quite rare.

## Experimental

A methanol solution ( 5 ml ) of $2-\mathrm{mBIM}(88 \mathrm{mg}, 0.5 \mathrm{mmol})$ was diffused slowly into an aqueous solution $(5 \mathrm{ml})$ of $\mathrm{Cd}\left(\mathrm{NO}_{3}\right)_{2} \cdot 4 \mathrm{H}_{2} \mathrm{O}$ $(154 \mathrm{mg}, 0.5 \mathrm{mmol})$ in a test tube. Colorless crystals of (I) were formed at the solvent interface after two weeks in $84 \%$ yield.

## Crystal data

$\left[\mathrm{Cd}\left(\mathrm{NO}_{3}\right)_{2}\left(\mathrm{C}_{9} \mathrm{H}_{12} \mathrm{~N}_{4}\right)_{2}\right]$
$\gamma=69.787$ (2) ${ }^{\circ}$
$M_{r}=588.87$
Triclinic, $P \overline{1}$
$a=7.9252$ (9) A
$b=9.2783$ (10) $\AA$
$c=9.4118$ (10) A
$\alpha=62.192$ (2) ${ }^{\circ}$
$\beta=73.725$ (2) ${ }^{\circ}$
Data collection
Bruker SMART APEX CCD areadetector diffractometer
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.74, T_{\text {max }}=0.78$

## Refinement

| $R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.031$ | 162 parameters |
| :--- | :--- |
| $w R\left(F^{2}\right)=0.068$ | H -atom parameters constrained |
| $S=1.07$ | $\Delta \rho_{\max }=0.77 \mathrm{e}^{-3}$ |
| 2229 reflections | $\Delta \rho_{\min }=-0.44 \mathrm{e}^{-3}$ |

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{Cd} 1-\mathrm{N} 2$ | $2.338(2)$ | $\mathrm{Cd} 1-\mathrm{O} 1$ | $2.404(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Cd} 1-\mathrm{N} 5^{\mathrm{ii}}$ | $2.383(2)$ |  |  |
| $\mathrm{N} 2^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{N} 5^{\mathrm{ii}}$ | $85.17(8)$ | $\mathrm{N} 2-\mathrm{Cd} 1-\mathrm{O} 1$ | $95.91(7)$ |
| $\mathrm{N} 2-\mathrm{Cd} 1-\mathrm{N} 5^{\mathrm{ii}}$ | $94.83(8)$ | $\mathrm{N} 5^{\mathrm{ii}}-\mathrm{Cd} 1-\mathrm{O} 1$ | $98.57(7)$ |
| $\mathrm{N} 2^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{O} 1$ | $84.09(7)$ | $\mathrm{N} 5^{\mathrm{iii}}-\mathrm{Cd} 1-\mathrm{O} 1$ | $81.43(7)$ |

Symmetry codes: (i) $-x+1,-y,-z+2$; (ii) $x, y-1, z+1$; (iii) $-x+1,-y+1$, $-z+1$.

H atoms were positioned geometrically at distances of $0.93(\mathrm{CH})$, $0.97\left(\mathrm{CH}_{2}\right)$ and $0.96 \AA\left(\mathrm{CH}_{3}\right)$ from the parent C atoms; a riding model was used during the refinement process. The $U_{\text {iso }}(\mathrm{H})$ values were constrained to be $1.2 U_{\text {eq }}$ of the carrier atom, except for methyl H atoms, which were constrained to be $1.5 U_{\text {eq }}$ of the parent C atom.

Data collection: SMART (Bruker, 2001); cell refinement: SAINTPlus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL.

We gratefully acknowledge the financial support of the National Science Funds for Distinguished Young Scholars of Hubei Province, People's Republic of China (grant No. 2006ABB038), the Outstanding Mid-young Scholars Programs, Hubei Provincial Department of Education (Q20072203), and the project sponsored by SRF for ROCS, SEM (200724).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SF3054). Services for accessing these data are described at the back of the journal.

## References

Bruker (2001). SAINT-Plus (Version 6.45), SMART (Version 5.628) and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Bünzli, J.-C. \& Piguet, C. (2002). Chem. Rev. 102, 1897-1928.
Carlucci, L., Ciani, G. \& Proserpio, D. (2003). Coord. Chem. Rev. 246, 247-289.
Dong, Y. B., Smith, M. D. \& zur Loye, H. C. (2000). Inorg. Chem. 39, 49274935.

Jin, C.-M. \& Gong, C.-R. (2005). Acta Cryst. E61, o1494-o1495.
Jin, C. M., Lu, H., Wu, L. Y. \& Huang, J. (2006). Chem. Commun. pp. 50395041.

Jung, O.-S., Kim, Y. J., Kim, K. M. \& Lee, Y.-A. (2002). J. Am. Chem. Soc. 124, 7906-7907.
Kasai, K., Aoyagi, M. \& Fujita, M. (2000). J. Am. Chem. Soc. 122, 2140-2141.
Moulton, B. \& Zaworotko, M. J. (2001). Chem. Rev. 101, 1629-1658.
Sharma, C. K. V. \& Rogers, R. D. (1999). Chem. Commun. pp. 83-84.
Sheldrick, G. M. (1996). $S A D A B S$. Version 2.10. Bruker AXS Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Wang, R., Hong, M., Weng, J., Su, W. \& Cao, R. (2000). Inorg. Chem. Commun. 3, 486-489.

